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#### Research article

## Influence of *Lysinibacillus sphaericus* on compressive strength and water sorptivity in microbial cement mortar



Daniel Karanja Mutitu<sup>a,\*</sup>, Jackson Muthengia Wachira<sup>a</sup>, Romano Mwirichia<sup>b</sup>, Joseph Karanja Thiong'o<sup>c</sup>, Onesmus Mulwa Munyao<sup>c</sup>, Genson Muriithi<sup>a</sup>

- <sup>a</sup> Department of Physical Sciences, University of Embu, Embu, Kenya
- <sup>b</sup> Department of Biological Sciences, University of Embu, Embu, Kenya
- <sup>c</sup> Department of Chemistry, Kenyatta University, Nairobi, Kenya

#### ARTICLE INFO

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#### ABSTRACT

Cement structures are subject to degradation either by aggressive media or development of micro/macro cracks which create external substance ingress pathways. Microbiocementation can be employed as a self-intelligent solution to this deterioration process. This paper presents study results on the effects of *Lysinibacillus sphaericus* microbiocementation on Ordinary Portland cement (OPC), normal consistency, setting time, soundness, compressive strength and water sorptivity. Microbial solutions with a concentration of  $1.0 \times 10^7$  cells/ml were mixed with OPC to make prisms at a water/cement ratio of 0.5. Mortar prisms of 160 mm  $\times$  40 mm x 40mm were used in this study. A maximum compressive strength gain of 17% and 19.8% was observed on the microbial prism at the  $28^{th}$  and  $56^{th}$  day of curing respectively. A minimum of 0.0190 and a maximum of 0.0355 water sorptivity coefficient was observed on the OPC microbial prism and OPC control prism, after  $28^{th}$  day of curing respectively. Scanning electron microscope images taken after the  $28^{th}$  day of curing showed formation of vast calcium silicate hydrates and more calcite deposits on microbial mortars. Statistical findings of this study indicate that *Lysinibacillus sphaericus* significantly retarded both the setting time and normal consistency, but has no influence on the mortar soundness.

#### 1. Introduction

Mortar/concrete is one of the most widely used construction material by mankind and it is the main material used for the infrastructure development of every country [1, 2]. Microorganisms in soils and waters play a major role on physico-mechanical properties and durability of building materials through the process generally referred to as microbial biocementation [3]. The process is affected by variation in temperatures, humidity, type of cement, type and concentration of bacteria and soil/water pH among others [4, 5]. Through microbial biocementation, calcium carbonate is deposited in cement mortar or concrete matrix. Such deposits have recently emerged as promising binders for protecting and consolidating various building materials [3, 6].

It is generally accepted that the durability of concrete/mortar is related to the characteristics of its pore structure [6]. Degradation mechanisms of concrete/mortar often depend on the way potentially aggressive substances can penetrate into the concrete/mortar, possibly causing damage [7]. The permeability of the concrete/mortar is

dependent on the porosity and on the connectivity of the pores [8, 9]. The more open the pore structure of the concrete/mortar, the more vulnerable the material is to degradation mechanisms caused by penetrating substances [1, 10].

OPC based structures are well known of achieving high early compressive and flexural strength large content and early hydration of C<sub>3</sub>S [1, 2, 11]. Despite the various OPC concrete/mortar advantages, it has more open pore structure than blended cements structures [1, 9, 12, 13]. This is one of the basic principles of adding blends such as pozzolana in cements to decrease these void spaces and consequently improve on porosity and connectivity of the pores. Addition of any blend/pozzolanic material during cement manufacturing process has a limitation to avoid compromising other beneficial characteristics of the clinker [14]. OPC mortar/concrete structures also have a high tendency to form cracks during and after curing allowing aggressive substances to penetrate into the structure. Permeability or cracks are one of the main causes of concrete/mortar deterioration and decrease in durability. Treatment of cracks and pores in concrete/mortar are generally divided into passive

E-mail address: mutsonjnr@gmail.com (D.K. Mutitu).

<sup>\*</sup> Corresponding author.

and active treatments. Passive treatments can only heal the surface cracks, while active treatments can heal both interior and exterior cracks [15, 16].

Microbial concrete/mortar biologically produce calcium carbonate (limestone) to seal pores that appear within the concrete/mortar matrix or heal cracks that appear on the surface of concrete/mortar structures [2, 16]. These microbrial deposits could also establish nucleation sites which enhance early cement hydration process leading to improved compressive and flexural strengths. Specific types of the bacteria genus such as Bacillus species along with a calcium-based nutrient such as calcium lactate, or calcium nitrate could be added to the ingredients of the concrete when it is being mixed [14]. These self-healing agents can lie dormant within the concrete/mortar for up to 200 years [15, 16]. However, when a concrete/mortar structure develops a crack or is damaged and water starts to seep through the cracks that appear in the concrete/mortar, the spores of the bacteria germinate on contact with the water and the nutrients. Once activated, the bacteria start to feed on the calcium-containing nutrient consuming oxygen. The calcium-containing nutrient is converted to insoluble calcium carbonate [9, 15, 16]. The calcium carbonate solidifies on the cracked surface, thereby sealing it up.

Oxygen is an essential element in the process of rebar corrosion and as such when the microbial activity consumes it, this improves the durability of steel reinforced concrete constructions. Thus, consumption of oxygen during the microbial conversion of calcium-containing nutrient to calcium carbonate has an additional advantage over sealing of pores/cracks [17]

Addition of bacterium in OPC mortar/concrete matrix is an innovative treatment which improve the transport/migration properties of OPC making these microbial treated concrete/mortars to act as intelligent systems that are different from the usually prepared OPC cement structures [13, 14]. These smart structures have self-sensing and self-healing properties towards external factors such as change in temperature, pH, humidity and concrete/mortar pore solution chemistry [5, 9, 16].

Microbiologically Induced Calcite Precipitation (MICP) reactions [18, 19, 20] can be summarized as:

$$CO(NH2)2 + H2O \rightarrow NH2COOH + NH3$$
 (1)

$$NH2COOH + H2O \rightarrow NH3 + H2CO3$$
 (2)

$$2NH_3 + 2H_2O \rightarrow 2NH_4^+ + 2OH^-$$
 (3)

$$H_2CO_3 \rightarrow HCO_3^- + H^+ \tag{4}$$

$$HCO_3^- + H^+ + 2OH^- \rightarrow CO_3^{2-} + 2H_2O$$
 (5)

$$Ca^{2+}$$
 + Bacterial cell  $\rightarrow$  Bacterial Cell-Ca<sup>2+</sup> (6)

Bacterial Cell-Ca<sup>2+</sup> + CO<sub>3</sub><sup>2-</sup> 
$$\rightarrow$$
 Cell-CaCO<sub>3</sub> (7)

#### 2. Materials and method

#### 2.1. Materials

#### 2.1.1. Cement chemical analysis

In this study, the Ordinary Portland Cement (OPC 42N) and standard sand manufactured according to ISO 679:1989, EN 196–1 [21] used were tested according to Kenyan Cement Standards specifications, KS EAS 18–1: 2017 [22]. 100 g sample of the test cement sample was ground to pass through a 76  $\mu m$  mesh sieve. The ground sample was used for chemical analysis of cement oxides using X-ray fluorescence in the usual manner [22]. Loss on Ignition was done in accordance to ASTM D7348: 2013 [23].

#### 2.1.2. Microbial culturing nutrients

Analytical grade (AR) chemicals were used in preparing the culture

media. Calcium lactate,  $C_6H_{10}O_6Ca$ , Peptone from casein and other animal proteins, Meat extract, Agar, sodium hydrogen carbonate (NaHCO<sub>3</sub>), anhydrous sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>), distilled water among other nutrients were purchased from Chem-Labs Limited, Nairobi, Kenya. *Lysinibacillus sphaericus* bacteria (DSM 28) was purchased from Leibniz-Institut DSMZ-Deutsche Sammlung von, Germany.

#### 2.2. Lysinibacillus sphaericus microbial culturing

The Lysinibacillus sphaericus microbial solution was cultured using nutrients as per the supplier manual. The liquid medium chosen for culturing the bacteria consisted of 5.00 g of peptone added to 3.00 g of meat extract and 3.95 g of calcium acetate per liter of distilled water was mixed to obtain liquid medium per stock culture. Initially, this mixture was sterilized for 20 min at a temperature of 121 °C by autoclaving. This mixture was then cooled to room temperature. After cooling, a 1M Nasesquicarbonate solution (1.0 ml in 10.0 ml) prepared by mixing 4.2 g NaHCO<sub>3</sub> with 5.3 g anhydrous Na<sub>2</sub>CO<sub>3</sub> and made up-to 1 L using distilled water was added to the stock culture to achieve a pH of 9.7. The Lysininbacillus sphaericus spore powder sample was added to this mixture in a laminar flow chamber. These cultures were then incubated on a shaker incubator at 130 rotations per minute maintained at 30 °C for 72 h. Optical density test was conducted using spectrophotometer for determining the quantity of culture solution required to mix. This test was conducted in bacteria growing medium which was considered as blank. This solution was also taken to be the reference, for experimentation of optical density of microbial solution. Separately, 0.5 mL of blank and bacteria solution of 0.5 mL were placed in the spectrophotometer at a wavelength of 600 nm and the machine set to read. The microbial concentration was observed to be  $1.0 \times 10^7$  cells/mL using the spectrophotometer. This microbial culture concentration was maintained throughout the mortar samples preparation as well as in prism curing solution.

#### 2.3. Mortar prism moulding and testing

Mortar mix prisms were fabricated according to KS EAS 18-1: 2017 [22]. 450 g of OPC was placed in the mix basin of an automatic programmable mixer model number JJ-5. 225.0 ml of distilled water was then added. The mix basin and its contents were clamped onto the automatic programmable mixer and allowed to run for 3 min 1350  $\pm$  5 g of the standard sand was placed in an automatic pour-trough little by little until all the 1350  $\pm$  5 g sample was added while the mixer was still running at a speed of 30 revolutions per minute (rpm). The machine was let to run for 10 min. The mortar prepared had w/c ratio of 0.50 and was sufficient to prepare three mortar prisms. Once the mortar was mixed, it was poured into steel moulds of 40 mm × 40 mm x 160 mm. Using a trowel, the mortar paste was scooped from the automatic programmable mixing basin and placed in a compaction mould of a jolting compaction machine with 60 rpm vibrations. Leveling of the paste was done with a mould trowel in each of the three chambers of the mould after every jolting cycle until a good finish was achieved at the surface. The mould with the mortar paste was then placed in a humid chamber maintained at 95% humidity and 27.0 °C for 24 h. The mortar was then demoulded from the moulds after 24 h to obtain usual OPC mortar. The distilled-water prepared mortars were categorized into two categories depending on their curing regime: The first category was cured in distilled water (labelled as OPC-H2O (H2O). The second category was cured in microbial solution (labelled as OPC-H2O [LB]). The above procedure was repeated but this time using 225 ml of microbial solution as mix media instead of distilled water which resulted to two more mortar categories: The third category was the OPC mortar prepared using microbial solution and cured in distilled water (labelled as OPC-LB [H<sub>2</sub>O]), while the forth category was the OPC mortar prepared using the microbial solution and cured in microbial solution (labelled as OPC-LB [LB]). The mortars were labeled for identification and placed in requisite water

or microbial solution for curing in a chamber maintained at  $27\pm1~^{\circ}C$  for curing. The compressive strength tests were conducted at the  $2^{nd},~7^{th},~14^{th},~28^{th}$  and  $56^{th}$  day of curing. Compressive strength tests in this study were performed on three samples per category for obtaining average results [24].

#### 2.4. Fresh paste tests preparation

Two categories of fresh cement paste for fresh paste tests; Normal consistency, Soundness and Setting time were separately prepared. The first category of the paste was prepared using distilled water as the mix media and was labelled OPC (H<sub>2</sub>O). The second category of fresh cement paste was prepared using microbial solution as the mix media and was labelled OPC (LB).

#### 2.4.1. Normal consistency test

A cement paste for OPC  $(H_2)$  weighing 300.0 g was prepared in accordance to IS 4031–4:1988 [25]. Standard consistency was calculated using Eq. (8). The weight of water for OPC  $(H_2O)$  fresh cement paste refers to weight of distilled water used while for OPC (LB) fresh cement paste category, the weight of water refers to the weight of microbial solution used to make the microbial fresh cement paste.

Standard consistency(%) = 
$$\frac{\text{Weight of water added}}{\text{Weight of cement}} \times 100$$
 (8)

Normal consistency tests in this study were performed on three samples per category for obtaining average results [24].

#### 2.4.2. Soundness test

The soundness test was done in accordance to KS EAS 148–3: 2017 [26]. A lightly oiled mould was placed on a lightly oiled glass sheet and filled with mortar paste formed by gauging cement with 0.78 times the distilled water to prepare OPC ( $\rm H_2O$ ) fresh cement paste. The same procedure was repeated but now with 0.78 times of the microbial solution instead of distilled water for OPC (LB) fresh cement paste to give a paste of standard consistency for each mortar category [25, 26]. Similar test was repeated on three fresh cement pastes samples per category to obtain averages [24].

#### 2.4.3. Setting time of cement

2.4.3.1. Initial setting time. A fresh sample of cement paste for each mortar category was prepared in accordance to KS EAS 148–3: 2017 [26]. This test was done on OPC (H<sub>2</sub>O) and also on OPC (LB) cement pastes. Initial setting time tests in this study were performed on three samples per cement paste category for obtaining average results [24].

#### 2.4.3.2 Final setting time.

A fresh sample of cement paste for each mortar category was prepared in accordance to KS EAS 148–3: 2017 [26]. This test was done on OPC ( $\rm H_2O$ ) and also on OPC (LB) cement pastes. Final setting time tests in this study were performed on three samples per cement paste category for obtaining average results [24].

#### 2.5. Sorptivity test

Sorptivity test was done on the four categories of mortar prisms OPC-H<sub>2</sub>O (H<sub>2</sub>O), OPC-H<sub>2</sub>O (LB), OPC-LB (H<sub>2</sub>O) and OPC-LB (LB) after the 28<sup>th</sup> day of curing. Sorptivity test was carried out following the method prescribed by Achal et al. (2016) [27]. To calculate the sorptivity coefficient, the 28<sup>th</sup> day cured mortar prism was dried at 100 °C in a ventilated oven. The mortar prism was then submerged into water at a height of 5 mm above the base of the mortar prism with the 40 mm  $\times$  40 mm side facing downward. At regular time intervals (15 min, 30 min; 1 h, 1.5 h, 3 h, 5 h, 8 h, 24 h, 72 h, 96 h, 120 h, 144 h, and 168 h), the mortar prism was

removed from the water and its new weight determined after drying the submerged surface with a clean wet towel. Immediately after the measurement, the mortar prism was then re-submerged again into water up to the 168<sup>th</sup> hour. Three samples for each mortar category were exposed to this test simultaneously and the triplicate results obtained for averaging. The sorptivity coefficient (k), was obtained by using Eq. (9):

$$\frac{Q}{A} = k t^{\frac{1}{2}}$$
 (9)

where, Q was the amount of water absorbed, A is the cross-section of the specimen that was in contact with water and t was the exposure time. On plotting a graph of Q/A against square root of time, values of k were determined graphically.

#### 3. Results and discussion

#### 3.1. Cement oxides

The results for the chemical analysis of cement oxides and LOI in percent by mass for the OPC test cement are given in Table 1.

The chemical analysis results shows that the test OPC met the minimum chemical composition requirements [22, 23]. Using Bogues formula [28], the average phase composition for the test OPC is 65.115  $\pm$  0.854%, 14.485  $\pm$  0.913%, 3.899  $\pm$  0.013% and 10.355  $\pm$  0.018% for C<sub>3</sub>S, C<sub>2</sub>S, C<sub>3</sub>A and C<sub>4</sub>AF respectively. These study results confirm that the test cement has the major cement phases that meets the Kenya Bureau of standards acceptable cement phases range [26].

### 3.2. Normal consistency, setting time and Soundness for Control and Microbial OPC

Table 2 gives the results for normal consistency, Setting time and Soundness for Control and Microbial OPC.

The OPC (LB) cement paste showed a significant difference ( $t_{calc}=0.00004729,\,p=0.05$ ) in normal consistency with OPC (LB) being statistically lower than the OPC (H<sub>2</sub>) cement paste as shown in Table 2. This was in agreement with the findings of Mutitu and co-authors [5, 29, 30]. The soundness of both OPC (LB) and OPC (H<sub>2</sub>) cement pastes exhibited no significant difference ( $t_{calc}=0.5,\,p=0.05$ ) and hence they were similar. The initial and final setting time for OPC (LB) cement paste were statistically lower than for OPC (H<sub>2</sub>) cement paste. The initial and final setting time for both OPC (H<sub>2</sub>) and OPC (LB) were significantly different with  $t_{calc}=0.0005288$  and  $t_{calc}=0.003881,\,p=0.05$  respectively. Thus, the addition of Lysinibacillus sphaericus microbial solution as mortar making mix media in the preparation of mortars significantly retards both the initial and final setting time as well the normal consistence, but has no significant influence on the mortar soundness.

#### 3.3. Scanning electron microscope (SEM) analysis

Fig. 1(a) to (d) shows SEM analysis for both control and microbial mortar prisms after  $28^{th}$  day of curing. The SEM images display the formation of calcium-silicate-hydrate, C–S–H, calcium carbonate precipitation, CaCO $_3$ , needle type ettringite, Ettr., and presence of portlandite/calcium hydroxide, CH.

As SEM images in Fig. 1 illustrate, the OPC- $H_2O$  ( $H_2O$ ) mortar had no visible calcium carbonate deposits. However, the microbial mortars OPC- $H_2O$  (LB), OPC-LB ( $H_2O$ ) and OPC-LB (LB) showed significant calcium carbonate precipitates. This could be attributed to the MICP deposits from the *Lysinibacillus sphaericus* either present in mix media or present in the cultured curing solution [5, 31, 32]. There a lot of observable differences between the SEMs including morphology of CSH which densifies from b to c to d.

This could be attributed to the calcium carbonate precipitation by Lysinibacillus sphaericus which is largely attributed to increase in

**Table 1**OPC chemical analysis results.

Sample	Cement Composition % w/w ±S.D									
	Al <sub>2</sub> O <sub>3</sub>	SiO <sub>2</sub>	SO <sub>3</sub>	Na <sub>2</sub> O	K <sub>2</sub> O	CaO	MgO	Fe <sub>2</sub> O <sub>3</sub>	MnO	LOI
Avg	3.643 ± 0.010	22.182 ± 0.010	$2.695 \pm 0.021$	0.410 ± 0.001	0.975 ± 0.006	$64.627 \pm 0.042$	$2.084 \pm 0.025$	3.403 ± 0.012	0.173 ± 0.001	1.519 ± 0.001

**Table 2**Normal consistency, Setting time and Soundness for Control and Microbial OPC.

Test Cement	Setting Ti	me (min)	Normal consistency	Soundness	
mortar	Initial Final		(%)	(mm)	
OPC (H <sub>2</sub> O)	98.0 ± 5.0	178.0 ± 5.0	$28.0\pm0.05$	$1.0\pm0.05$	
OPC (LB)	$78.0\ \pm$ $5.0$	$167.0 \pm 5.0$	$26.4\pm0.05$	$1.0 \pm 0.05$	

compressive strength as well decrease in sorptivity coefficient. Image (c) from Fig. 1, clearly shows biodeposition over ettringite needles resulting to formation of biofilms on their surface and plugging of the pores on the mortar structure. This could explain why MICP technique has been used to remove sulphate and clean crusts from marble monuments [33, 34, 35].

#### 3.4. Compressive strength

The compressive strength results obtained at 14<sup>th</sup>, 28<sup>th</sup> and 56<sup>th</sup> day of curing are summarized in Fig. 2.

Compressive strength for OPC- $H_2O$  ( $H_2O$ ), OPC- $H_2O$  (LB), OPC-LB ( $H_2O$ ) and OPC-LB (LB) mortar prisms were determined using a compressive strength machine at the  $2^{nd}$ ,  $7^{th}$ ,  $14^{th}$ ,  $28^{th}$  and  $56^{th}$  day of curing. Across all mortar categories, there was no significant difference in their compressive strengths at  $2^{nd}$  and  $7^{th}$  day of curing and hence they have not been reported in this study. Perhaps the bacteria had not precipitated significant quantities of calcium carbonate or the cement material had not reacted with the precipitation to form a strength beneficial material. A summary of significance difference in compressive strength among different mortar categories have been presented in Table 3.

The compressive strength for all mortar categories increased with increase in curing age as depicted on Fig. 3. Across all curing ages, OPC-LB (LB) exhibited the highest compressive strength as well the highest percent gain in compressive strength than the other mortar categories. The highest compressive strength and percent compressive strength gain was observed at the 56<sup>th</sup> day of curing at 58.6MPa and 19.8% as shown on Fig. 2 and Fig. 3 respectively. There was observed a statistically significant difference in compressive strength and percent compressive strength gain both from one curing age to another as well from one microbial mortar category to another for all microbial mortar categories. as

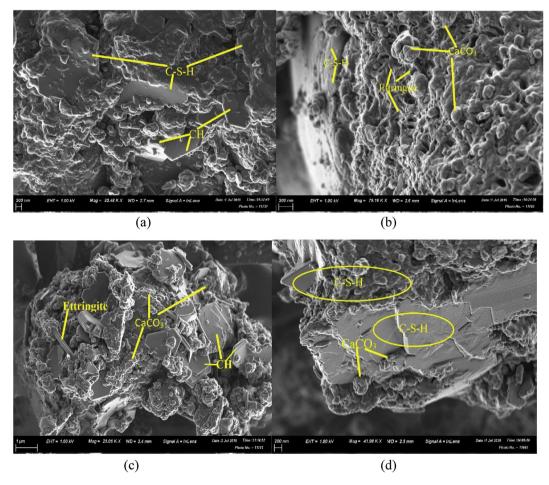


Fig. 1. SEM analysis for (a) OPC-H<sub>2</sub>O (H<sub>2</sub>O), (b) OPC-H<sub>2</sub>O (LB), (c) OPC-LB (H<sub>2</sub>O) and (d) OPC-LB (LB).

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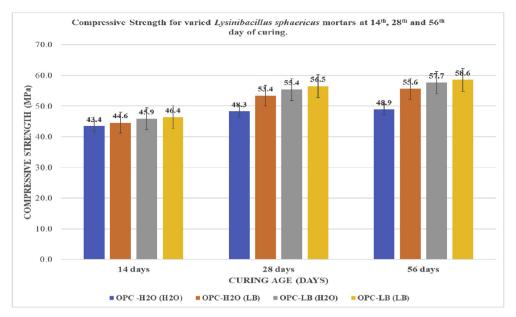


Fig. 2. Results for Compressive Strength of test mortars at 14th, 28th and 56th day of Curing.

**Table 3**  $T_{Calc.}$  values summary for mortar categories compressive strength across  $2^{nd}$ ,  $7^{th}$ ,  $14^{th}$ ,  $28^{th}$  and  $56^{th}$  day of curing. ( $T_{Crit.}=0.5$ , p=0.05).

MORTAR CATEGORIES	$T_{Calc}$ . Values (x10 <sup>-3</sup> )						
	2 <sup>nd</sup> day	7 <sup>th</sup> day	14 <sup>th</sup> day	28 <sup>th</sup> day	56 <sup>th</sup> day		
OPC-H <sub>2</sub> O (H <sub>2</sub> O) vs. OPC- H <sub>2</sub> O (LB)	500	500	1.4271	0.0012	0.0270		
OPC-H2O ( $H_2O$ ) vs. OPC-LB ( $H_2O$ )	500	500	0.5483	0.0001	0.0002		
OPC-H <sub>2</sub> O (H <sub>2</sub> O) vs. OPC-LB (LB)	500	500	0.0254	0.0332	0.2756		
OPC-H <sub>2</sub> O (LB) vs. OPC-LB (H <sub>2</sub> O)	500	500	0.1061	0.0262	0.1222		
OPC-H <sub>2</sub> O (LB) vs. OPC-LB (LB)	500	500	0.0833	0.0905	1.4317		
OPC-LB (H <sub>2</sub> O) vs. OPC-LB (LB)	500	500	6.2184	3.2996	31.7364		

shown in Table 2. The increase in compressive strength could be attributed to the materials precipitated by the *Lysinibacillus sphaericus* being involved in the hydration process forming calcium silicate hydrate responsible for strength development. The added  $Ca^{2+}$  together with calcium acetate, in presence of the microbial cell-wall nucleation site readily combine with the precipitated  $CO_3^{2-}$  and crystallizes out as calcium silicate hydrate (C–S–H) [5, 16, 27, 30].

In this study, it has been found that the *Lysinibacillus sphaericus* biomineralization process enhance the compressive strength hence improving durability properties of OPC cement structures. The enhanced MICP process in this study, could also be attributed to the metabolic conversion of the organic acetate added as a microbial feed in form of calcium acetate which was aerobically oxidized under improved alkaline conditions by this ureolytic alkaliphilic *Bacillus spp.* according to the following equation [5, 16]:

$$CH_3COO^{1-} + 2O_2 \rightarrow 2CO_2 + OH^{1-} + H_2O$$
 (10)

$$CO_2 + OH^{1-} \rightarrow HCO_3^{1-}$$
 (11)

$$HCO_3^{1-} + OH^{1-} \rightarrow CO_3^{2-} + H_2O$$
 (12)

Bacterial Cell-Ca<sup>2+</sup> + CO<sub>3</sub><sup>2-</sup> 
$$\rightarrow$$
 Cell-CaCO<sub>3</sub> (13)

#### 3.5. Water sorptivity

As shown in Fig. 4, throughout the water exposure duration, OPC- $\rm H_2O$  ( $\rm H_2O$ ), exhibited the highest percent water absorption while the OPC-LB (LB), had the lowest. All the test mortars exhibited a gradual rise in water absorption but after the  $\rm 120^{th}$  hour of water exposure the water sorption almost became constant. Perhaps the mortars became water saturated such that they could not absorb more water.

As shown on Fig. 5 and summarized in Fig. 6, it is evident that all the microbial mortar prisms OPC-H<sub>2</sub>O (LB), OPC-LB (H<sub>2</sub>O) and OPC-LB (LB) had a higher water absorption resistance than the non-microbial OPC-H<sub>2</sub>O (H<sub>2</sub>O) mortar prism. The decrease in water sorption could be attributed to the calcium carbonate precipitation which is also evident in the SEM images as shown in Fig. 1 (a) as compared to (b), (c) and (d). Other researchers though using different bacteria such as Mutitu and his co-authors [5], Achal and his co-workers [27] using Sporosarcina pasteurii and Dhamia and his co-workers [36] using Bacillus megaterium bacteria also observed water absorption reduction in the microbial mortars as compared with mortars prepared without microbial media. All these studies attributed the decrease in water permeability and porosity to the calcium carbonate precipitation which seals the pores in the mortar matrix inhibiting water migration [5, 27, 37].

#### 4. Conclusion

- Lysinibacillus sphaericus has the ability to precipitation-controlled amount of calcium carbonate, enough to improve both compressive strengths as well the pore structure lowering water ingress.
- Compressive strength of the mortar increased with increase in the population/concentration of *Lysinibacillus sphaericus* bacteria. This was exhibited by the higher strengths in microbial mortars cured in cultured solutions showing higher strengths than those cured in distilled water.
- Incorporation of Lysinibacillus sphaericus into OPC mortar enhanced the nucleation sites for calcite precipitation leading to improved overall microstructure of the mortar. The calcite precipitation filled the pores of the mortar reducing water permeation.
- 4. Incorporation of calcium acetate as a feed in the microbial culturing introduced more Ca<sup>2+</sup> in the mortar matrix. This enhances formation of the cement-compatible calcium carbonate which plugs the pores as

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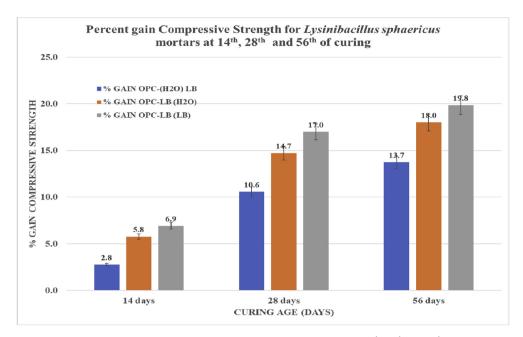


Fig. 3. Percent gain Compressive strength of Lysinibacillus sphaericus mortars at 14th, 28th and 56th day of curing.

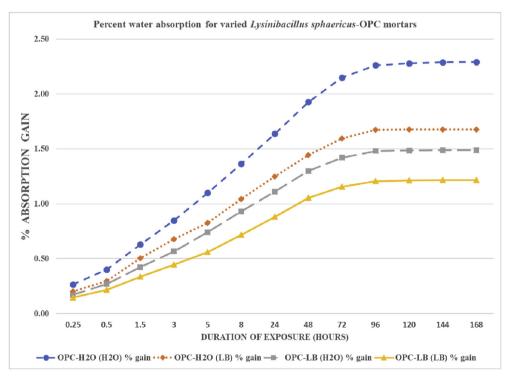


Fig. 4. Percent water absorption for varied Lysinibacillus sphaericus OPC mortars after 28th day of curing.

shown by the SEM images improving both the compressive strength as well lowering water sorptivity.

#### Declarations

#### Author contribution statement

Daniel Karanja Mutitu: Performed the experiments; Analyzed and interpreted the data; Wrote the paper.

Jackson Muthengia Wachira, Romano Mwirichia, Joseph Karanja Thiong'o, Onesmus Mulwa Munyao, Genson Muriithi: Conceived and

designed the experiments; Contributed reagents, materials, analysis tools or data.

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#### Competing interest statement

The authors declare no conflict of interest.

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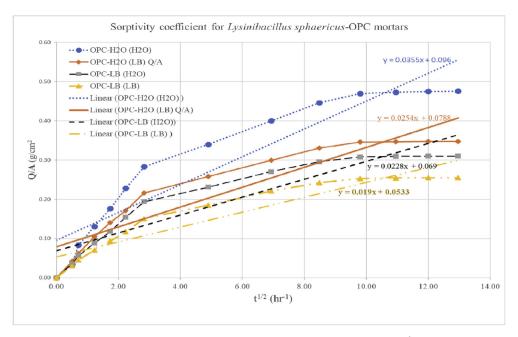


Fig. 5. Sorptivity coefficients for varied Lysinibacillus sphaericus-OPC mortars after 28th day of curing.

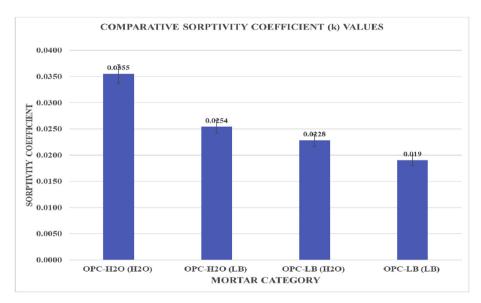


Fig. 6. Comparative Sorptivity coefficients for varied Lysinibacillus sphaericus mortars after 28th day of curing.

#### Additional information

No additional information is available for this paper.

#### Data availability statement

The raw or any additional data used to support the findings of this study are restricted by the University of Embu, Board of Postgraduate in order to avoid plagiarism since this study findings are intended to form part of a PhD thesis yet to be published.

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